



A rhodium(I)-xylyl-BINAP catalyzed asymmetric ynamide-[2+2+2] cycloaddition in the synthesis of optically enriched *N,O*-biaryls

Jossian Oppenheimer, Whitney L. Johnson, Ruth Figueroa, Ryuji Hayashi, Richard P. Hsung*

Division of Pharmaceutical Sciences and Department of Chemistry, University of Wisconsin, Madison, WI 53705, USA

ARTICLE INFO

Article history:

Received 30 January 2009
Received in revised form 23 March 2009
Accepted 26 March 2009
Available online 1 April 2009

Keywords:

Aryl ynamides
Asymmetric [2+2+2] cycloaddition
Chiral *N,O*-biaryls
Rhodium(I) catalyst and (*S*)-xylyl-BINAP
C–N axial chirality
double asymmetric induction of
non-point stereocenters

ABSTRACT

A rhodium(I)-xylyl-BINAP catalyzed asymmetric [2+2+2] cycloaddition of achiral conjugated aryl ynamides with various diynes is described here. This asymmetric cycloaddition provides a series of structurally interesting chiral *N,O*-biaryls with excellent enantioselectivity along with a modest diastereoselectivity with respect to both C–C and C–N axial chirality.

© 2009 Elsevier Ltd. All rights reserved.

1. Introduction

In the last 15 years, ynamides have captured much interest from the synthetic community. An immense amount of efforts has led to a number of elegant synthetic methodologies to be developed adopting ynamides as a de novo functional group,^{1–4} and culminated in several total syntheses employing ynamides as a versatile building block.^{5,6} Among these efforts [Fig. 1], both Witulski's work on [2+2+2] cycloadditions of ynamides in the synthesis of indoles and carbazoles,^{7,8} and Rainier's formal [2+2+2] cycloaddition⁹ inspired us^{10,11} to develop a new approach toward chiral *N,O*-biaryls through a [2+2+2] cycloaddition of conjugated aryl ynamides with diynes.^{12–14} This concept was envisioned at the heel of our success in the synthesis of conjugated aryl ynamides via Sonogashira cross-coupling.¹⁵ During our pursuit, Tanaka^{16,17} beautifully demonstrated the feasibility of an asymmetric [2+2+2] cycloaddition of ynamides in the synthesis of anilides with an axially chiral C–N bond.^{18–20}

Because our diastereoselective ynamide-[2+2+2] cycloaddition only resulted in modest selectivity through the use of chiral aryl ynamide **1** [Scheme 1],¹⁰ we began contemplating the possibility of pursuing an asymmetric cycloaddition employing achiral aryl ynamides **3** and a suitable chiral rhodium(I) complex.¹¹ If successful, asymmetry could be induced at both axially chiral C–C biaryl

bond [in blue] and C–N anilide bond [in red] in cycloadducts **5**, thereby constituting a rare example of double asymmetric induction of non-point stereocenters by a single catalyst.^{12,14d,20–22} Moreover, the coordinating ability of the achiral amido-carbonyl group could play a role in the asymmetric induction, thereby rendering this design an achiral-template directed asymmetric catalysis.²³ We report here our development of an asymmetric ynamide-[2+2+2] cycloaddition.

2. Results and discussions

2.1. Establishing the feasibility

In our earlier work, we were intrigued by the observation that while we could not improve the diastereoselectivity in our RhCl(Ph₃P)₃-catalyzed [2+2+2] cycloadditions of chiral ynamide **1**, we could readily diminish the ratio from 4:1 by using an external phosphine ligand such as BINAP [Scheme 2]. To account for the modest diastereoselectivity, we had proposed *pro-M*- and *pro-P*-TS models, which consist of the key bidentate coordination [in green] of Rh(III) from both the oxazolidinone carbonyl oxygen and the anisole oxygen atom. We concluded that neither really possesses a distinct advantage over the other except with *pro-P*-TS suffering from a remote steric interaction [red arrows] between the cyclopentyl and the phenyl group on the oxazolidinone ring. Based on these same models, the above finding suggests that an external chiral ligand [in blue] exerts a stronger influence on this

* Corresponding author. Tel.: +1 608 890 1063; fax: +1 608 262 5345.
E-mail address: rhsung@wisc.edu (R.P. Hsung).

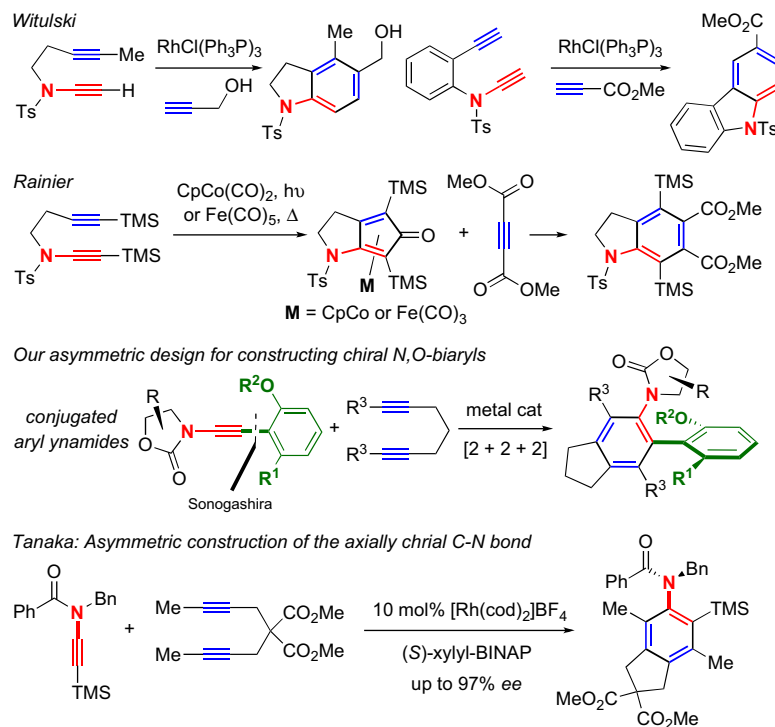
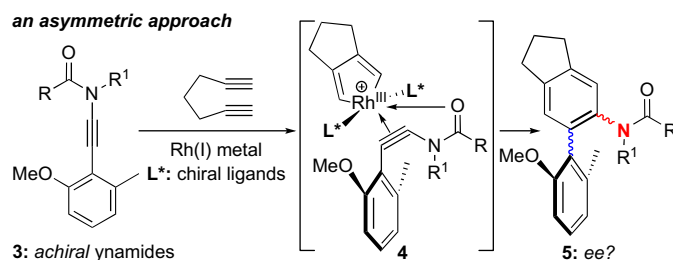
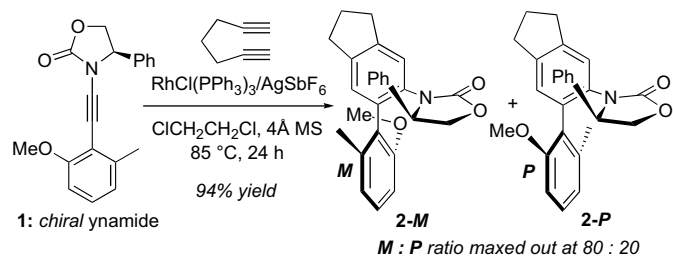


Figure 1. Ynamide-[2+2+2] cycloadditions.



Scheme 1. An asymmetric ynamide-[2+2+2] cycloaddition.

cycloaddition than the stereocenter on the chiral ynamide, thereby implying the possibility of an asymmetric cycloaddition manifold employing achiral ynamides.

Toward this goal, we synthesized achiral conjugated aryl ynamides **6** and **7**^{24–26} containing a sulfonamide and an acyclic amide, respectively, as shown in Scheme 3. However, to our disappointment, after screening several catalytic systems, we were unable to even find the desired cycloaddition products **8** and **9** let alone analyzing possible enantioselectivity. Recognizing that ynamides with the amido group that are cyclic in nature may be better suited for the cycloaddition, we proceeded to prepare aryl ynamide **10a** containing the 2-oxazolidinone ring [Scheme 4].

After screening again conditions such as $\text{Rh}(\text{PPh}_3)_3\text{Cl}/\text{AgSbF}_6$, $[\text{Rh}(\text{CO})_2\text{Cl}]_2$, $[\text{Rh}(\text{cod})\text{Cl}]_2/\text{AgSbF}_6$, and $[\text{Rh}(\text{cod})_2]\text{BF}_4$ activated

with H_2 , we found that the usage of 10 mol% $[\text{Rh}(\text{cod})_2]\text{BF}_4$ and 10 mol% (*S*)-xylyl-BINAP at 85 °C in $\text{ClCH}_2\text{CH}_2\text{Cl}$ with 4 Å MS provided the best outcome when reacting **10a** with diyne **11**. While the resulting diastereomeric *N,O*-biaryls **12-M,p** and **12-P,p**²⁷ were separable by TLC, they rapidly inter-converted to the enantiomer of each other: **12-M,p** led to **12-M,m** with **12-P,p** yielding **12-P,m** via C–N bond rotation [in red]. Even after a clean and facile separation on a silica gel column, NMR analysis revealed that the two diastereomers of **12** have already scrambled likely during the removal of the solvent under reduced pressure even without heating. Thus, this complication prevented us from meaningfully determining dr and ee.

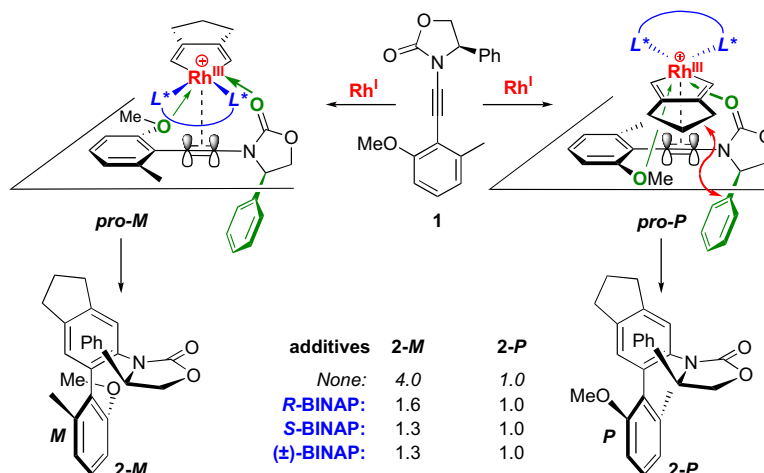
Ultimately, cycloadditions of achiral conjugated aryl ynamides **10b** and **10c** allowed us to analyze possible enantioselectivity. As shown in Scheme 5, *N,O*-biaryl atropisomers **14-M,p** and **14-P,p** were attained in 95% yield from **10b** when reacting with diyne **13a**. Although potential diastereomers of **14** due to restricted C–N bond rotation were detectable in proton NMR, they were not separable physically. Consequently, we were able to cleanly assess the enantiomeric excess using Chiral HPLC.

Furthermore, aryl ynamide **10c** containing the six-membered 2-oxazinone ring also led to *N,O*-biaryls **15** and **16** when reacting with diynes **13a** and **13b**, respectively. There was even less complication in these two reactions, as only a single diastereomer was detectable in ¹H NMR presumably due to rapid free rotation at the C–N bond. Thus, we were able to concisely calibrate ee values for both **15** and **16**, which are lower than that of **14**. It is noteworthy that since BINAP had given lower ee and yield than xylyl-BINAP, and with Tanaka's success using xylyl-BINAP, we did not screen beyond these two chiral bidentate phosphines.

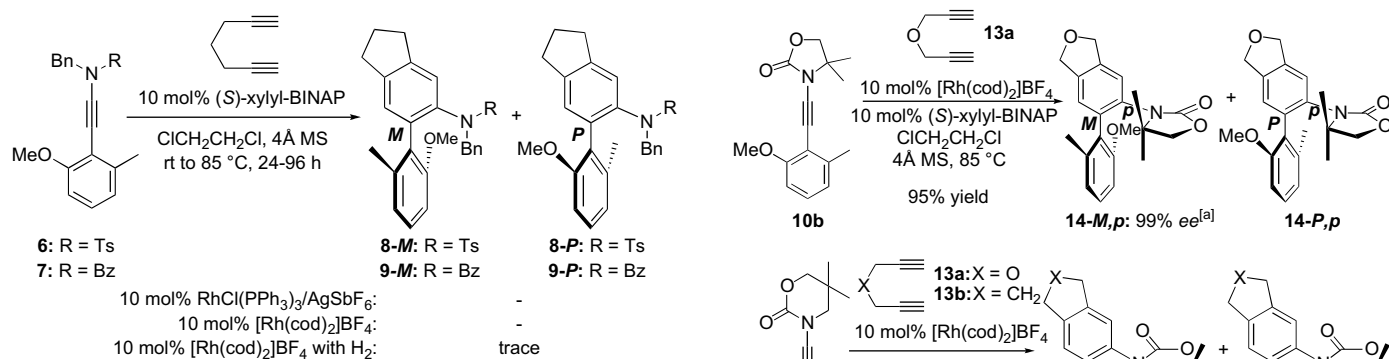
2.2. Syntheses of chiral *N,O*-biaryls

Having established an asymmetric protocol, a range of chiral *N,O*-biaryls could be prepared as shown in Figure 1. Reactions of aryl ynamide **10a** with diyne **11** led to the respective diastereomeric

an external ligand controlled process

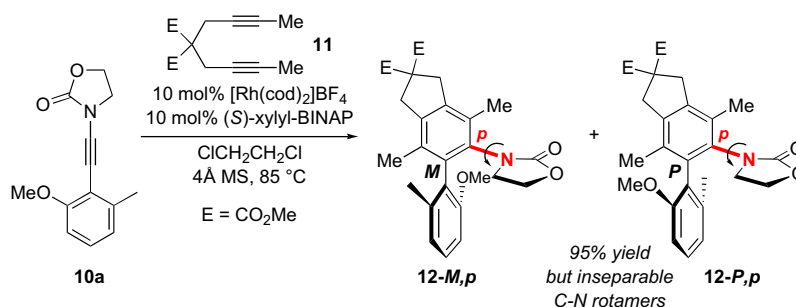


Scheme 2. Recognition of an external ligand controlled process.

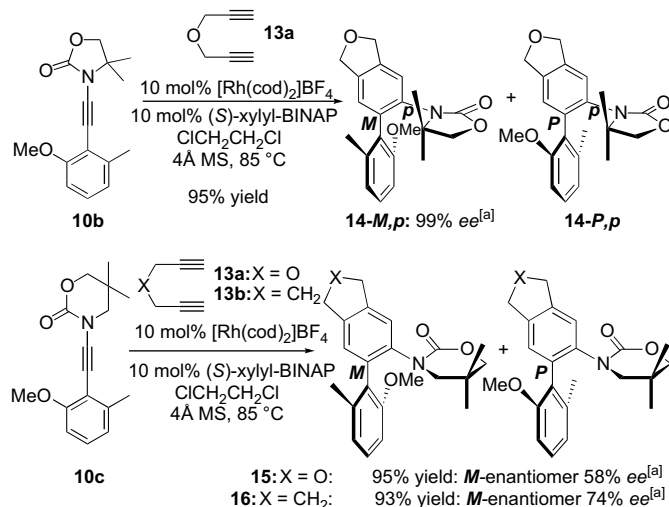


Scheme 3. Cycloadditions of ynamides with an acyclic amido-motif.

N,O-biaryl isomers **17** and **18** in good yields. Notably, we observed a diastereoselectivity as high as 8:1 in favor of the isomer **17-M,p**, which also possesses an enantiomeric excess of 95%, while the minor isomer **18-P,p** has a 54% ee. The usage of (*R*)-xylyl-BINAP led to chiral *N,O*-biaryls *ent*-**17** and *ent*-**18** but in slightly lower ee and dr. Moreover, reactions of ynamide **10b** containing the five-membered 2-oxazolidinone ring led to *N,O*-biaryls generally with higher ee for both *P,p* and *M,p* diastereomers. For example, while diastereomeric isomers **23-M,p** and **24-P,p** were attained with a dr of 6:1, both diastereomers possess 99% ee. Finally, while the ratio dropped for diastereomeric *N,O*-biaryls **27** and **28**, as well as for **29** and **30**, which all contain a naphthyl ring, their respective enantioselectivity remained high [Fig. 2].



Scheme 4. Cycloadditions of ynamides with a cyclic amido-motif.



Scheme 5. An asymmetric ynamide-[2+2+2] cycloaddition. ^aThe ee was determined using chiral HPLC [CHIRALCEL OD-H; size: 250×4.6 mm (L×i.d.); eluent: *i*-PrOH in hexanes].

2.3. Assignment of absolute configurations

The minor diastereomer **18-P,p** is crystalline, and its absolute configuration could be readily resolved through X-ray structure of a single crystal [Fig. 3].²⁸ However, assignments of the major isomer were not as trivial. As shown in Scheme 6, absolute configurations of major isomers **17-M,p** containing the six-membered 2-oxazinone

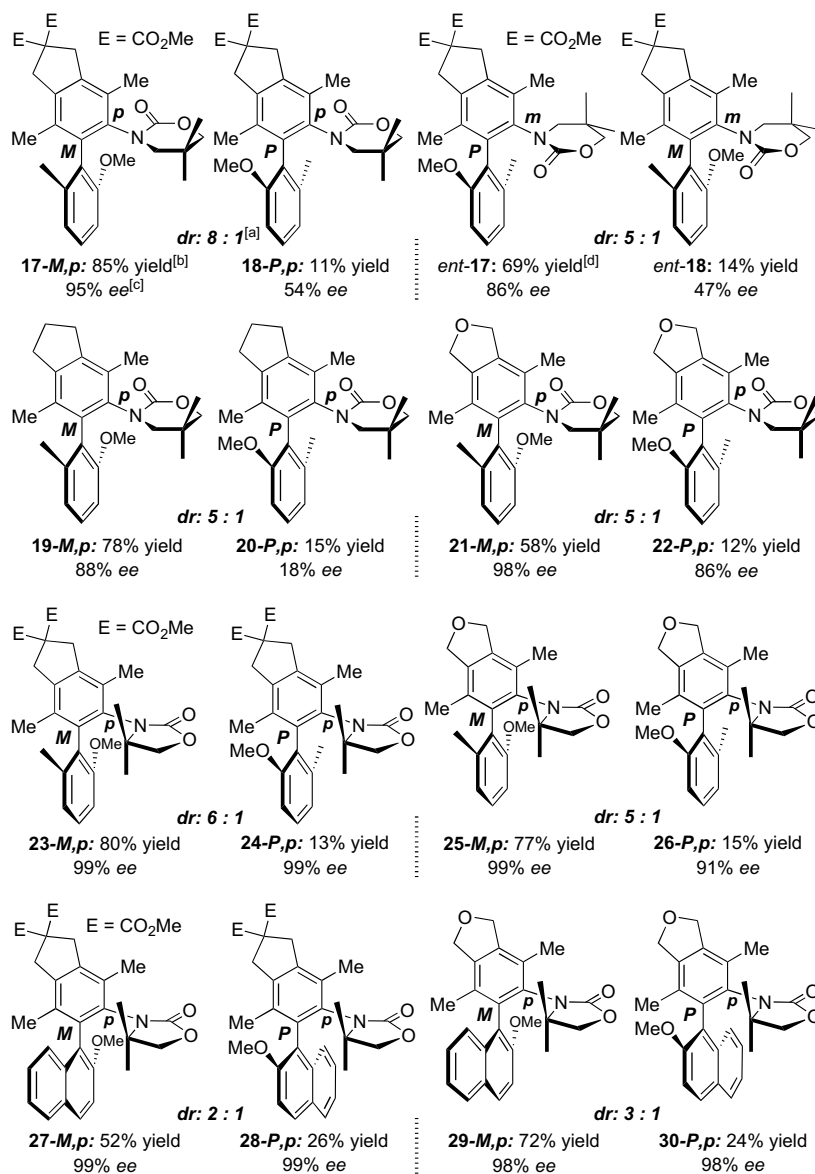


Figure 2. Scope of Asymmetric Ynamide-[2+2+2] Cycloadditions. ^aThe dr was determined using ¹H/¹³C NMR. ^bIsolated yields. ^cThe ee was determined using chiral HPLC [CHIRALCEL OD-H; size: 250×4.6 mm]; eluent: *i*-PrOH in hexanes. ^d(*R*)-xylyl-BINAP was used.

ring, and **23-M,p** containing the five-membered 2-oxazolidinone ring, had to be assigned via X-ray structures of their respective camphor-sulfonyl derivatives **32-M,p** and **34-M,p** prepared in two steps. X-ray structures of **32-M,p** and **34-M,p** are shown in Figure 4. Correlations of aromatic protons on the anisyl ring allow for the assignment of all other isomeric *N,O*-biaryls.

2.4. Synthesis of a chiral amino-biaryl

To demonstrate that these chiral *N,O*-biaryls can be useful in the venue of designing new chiral *N,O*-biaryl ligands, we pursued the following asymmetric cycloaddition. As shown in Scheme 7, cycloaddition of ynamide **35** containing the 2-oxazolidinone ring with *gem*-diphenyl substitutions^{11,25} with diene **36** gave cycloadducts **37-M,p** and **38-P,p** in 1:1 ratio with 82% ee for each diastereomer. We took **37-M,p** and removed the achiral 2-oxazolidinone ring via hydrogenations to afford aniline **39-M**. Although we were unable to clearly discern the final enantiomeric excess of **39-M** through chiral HPLC, with the C–C bond

rotational barrier being 37.7 kcal mol⁻¹ [PM3 calculations], we believe there should be very little racemization during the hydrogenation at rt.

2.5. Equilibration studies

We were fascinated with the unique structural feature of these *de novo* *N,O*-biaryls. Spartan[™] B3LYP/6-31G* calculations revealed a ΔE of 1.11 kcal mol⁻¹ in favor of the minor isomer **18**, thereby implying that the observed selectivity for **17** is kinetic. In addition, B3LYP/6-31G* calculations provided E_{act} of 34.0 kcal mol⁻¹ and 94.4 kcal mol⁻¹, respectively, for the axially chiral C–N and C–C bonds. These calculations would suggest that any thermal equilibrations would first lead to epimerization through the C–N rotation. That is the equilibration between **17-M,p** and **ent-18-M,m**, or between **18-P,p** and **ent-17-P,m** [see Scheme 8], should occur faster at lower temperature than that between **17-M,p** and **18-P,p**, or between **ent-18-M,m** and **ent-17-P,m**. To confirm these assertions, we carried out the following equilibration studies.

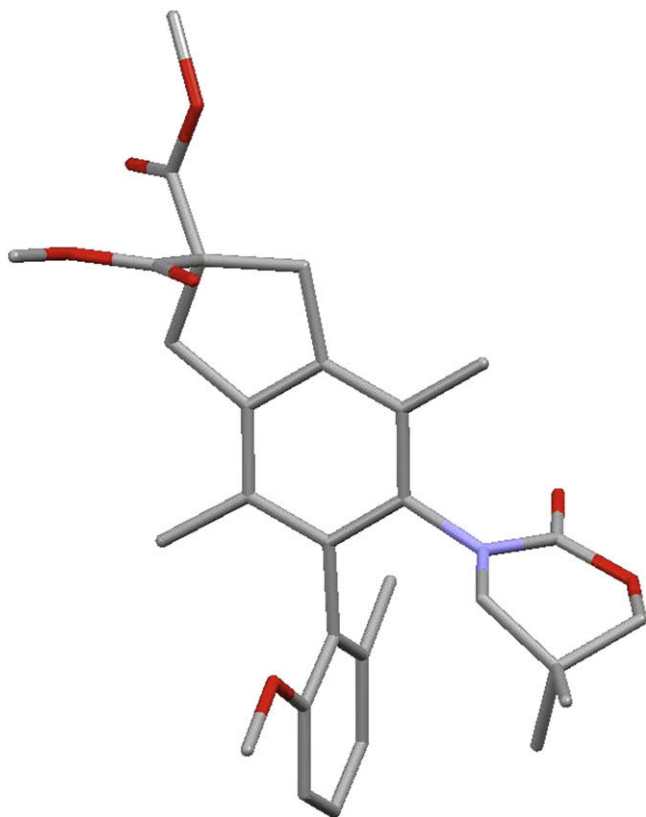


Figure 3. X-ray structures minor atropisomer **18-P,p**.

As shown in Table 1, heating pure biaryl **17-M,p** with a 95% ee at 85 °C in toluene-*d*₈ for 24 h [the original reaction temperature] did not result in any epimerization or loss of optical integrity. At 120 °C, epimerization occurred and led to a significant loss of **17-M,p** and an increase in the amount of *ent*-**18-M,m**. However, there was no loss of enantiomeric excess [94% ee] for both **17-M,p** and *ent*-**18-M,m**, thereby suggesting that at 120 °C, the epimerization was occurring solely through rotation of the axially chiral C–N bond [assuming enantiomers in general possess the same rate of epimerization in first order]. At 165 °C and 200 °C, while **17-M,p**

continued to epimerize to *ent*-**18-M,m** with the ratio gravitating toward 23:77, there was a noticeable loss in the ee for both **17-M,p** and *ent*-**18-M,m**. Thus, at these high temperatures, epimerization was taking place through free rotations at both C–N and C–C bonds, which should then lead to racemization.

Equilibrations of pure *ent*-**17-P,m** with an 86% ee led to a very similar outcome and likewise with equilibrations of **18-P,p** even starting at a 52% ee [Table 2]. It is noteworthy that all the equilibration studies led to a final resting thermodynamic ratio of 22:78 for **17/18**. This appears to be the thermodynamic ratio, which is consistent with the calculation in which isomer **18** is favored by 1.11 kcal mol⁻¹.

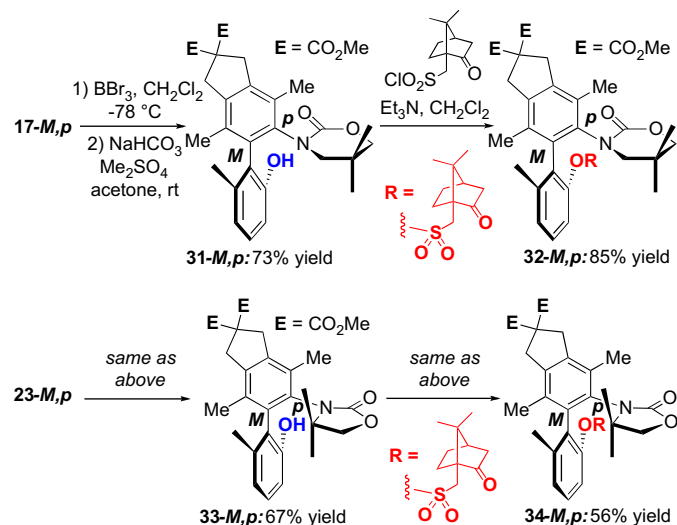
2.6. Mechanistic considerations

Given the stereochemical outcome, we proposed a mechanistic model that could provide a unified rationale for the observed asymmetric inductions. As shown in Scheme 9, with the rhodium-cyclopentadiene intermediate complexed to (*S*)-xylyl-BINAP, a respective ynamide could approach the metal more favorably as shown in complex **A1** with the anisole ring sliding into the less hindered space near **Ar**² with a smaller amido group assuming the relatively more crowded space next to **Ar**¹. This binding orientation would be the opposite in complex **B1** and would lead to a less favorable fit in the two respective ‘binding pockets’.

With complex **A1** in hand, a key bidentate chelation of the Rh metal via both the anisyl OMe and the carbonyl group could occur to assist the complexation of the ynamide unto the metal center. The ensuing cycloaddition, in a stepwise or Diels–Alder manner, would lead to the major diastereomer [*M,p*]. Because the enantiomer of the major isomer would come from complex **B1**, which is not favored in these reactions, enantiomeric excess is high in almost all cases for atropisomer [*M,p*].

If the bidentate chelation fails due to the slipping of the weaker coordination from the anisole oxygen atom through a C–C rotation, one would obtain complex **A2**, which would give the observed minor diastereomer [*P,p*] after the cycloaddition. On the other hand, if the C–N bond in **A1** rotates and one loses the stronger coordination from the carbonyl oxygen, complex **A3** would be present to give the corresponding [*M,m*] isomer, which is the enantiomer of the minor diastereomer. Since the carbonyl oxygen in general possesses better coordinating ability than a phenolic etheral oxygen, it is reasonable to see much less formation of [*M,m*], thereby rendering [*P,p*] as the major enantiomer for the minor diastereomer.

Finally, the importance of the carbonyl oxygen in this asymmetric cycloaddition can also be further underscored from the fact that 2-oxazolidinone is better chelator than 2-oxazinone. Consequently, ynamides substituted with 2-oxazolidinone gave higher ee. Overall, the current mechanistic model provides another excellent example for showcasing the versatility of enantioselective catalysis through an asymmetric template that employs an achiral auxiliary.²³



Scheme 6. Syntheses of **32-M,p** and **34-M,p**.

3. Conclusion

We have described here a rhodium(I)-xylyl-BINAP catalyzed asymmetric [2+2+2] cycloaddition of achiral conjugated aryl ynamides with various diynes. This asymmetric cycloaddition represents a rare example of double asymmetric induction of non-point stereocenters by a single catalyst and provides a series of structurally interesting chiral *N,O*-biaryls with excellent enantioselectivity in most cases along with a modest diastereomeric selectivity with respect to both C–C and C–N axial chirality.

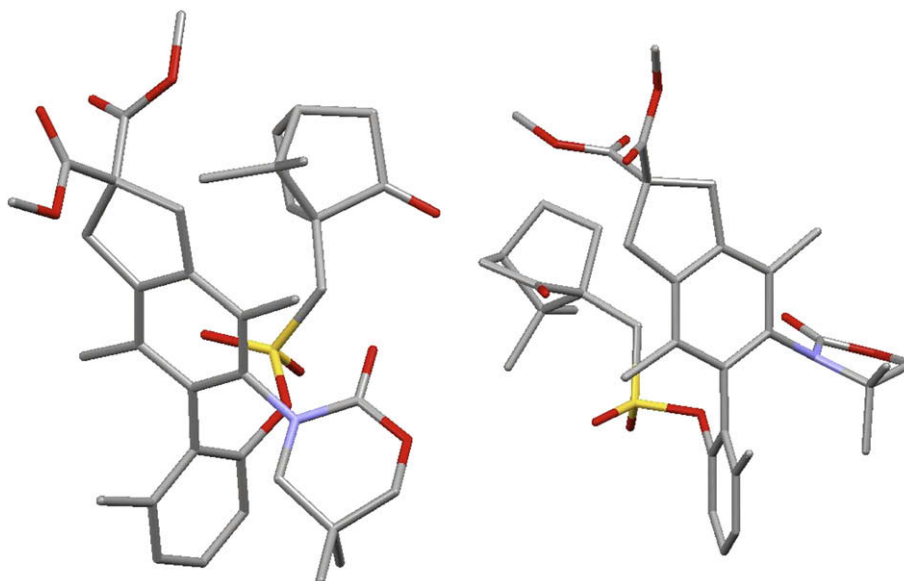


Figure 4. X-ray structures: **18-P,p 32-M,p** [left] and **34-M,p** [right].

4. Experimental section

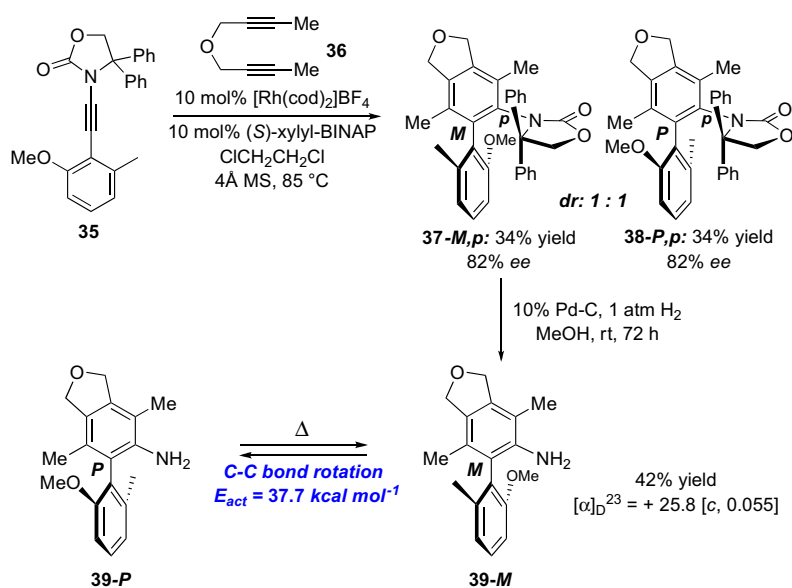
4.1. General

All reactions were performed in flame-dried glassware under a nitrogen atmosphere. Solvents were distilled prior to use. Reagents were used as purchased (Aldrich, Acros), except where noted. Chromatographic separations were performed using Bodman 60 Å SiO₂. ¹H and ¹³C NMR spectra were obtained on Varian VI-300, VI-400, and VI-500 spectrometers using CDCl₃ (except where noted) with TMS or residual CHCl₃ as standard. Melting points were determined using a Laboratory Devices MEL-TEMP and are uncorrected/calibrated. Infrared spectra were obtained using NaCl plates on a Bruker Equinox 55/S FT-IR Spectrophotometer, and relative intensities are expressed qualitatively as s (strong), m (medium), and w (weak). TLC analysis was performed using Aldrich 254 nm polyester-backed plates (60 Å, 250 μm) and visualized

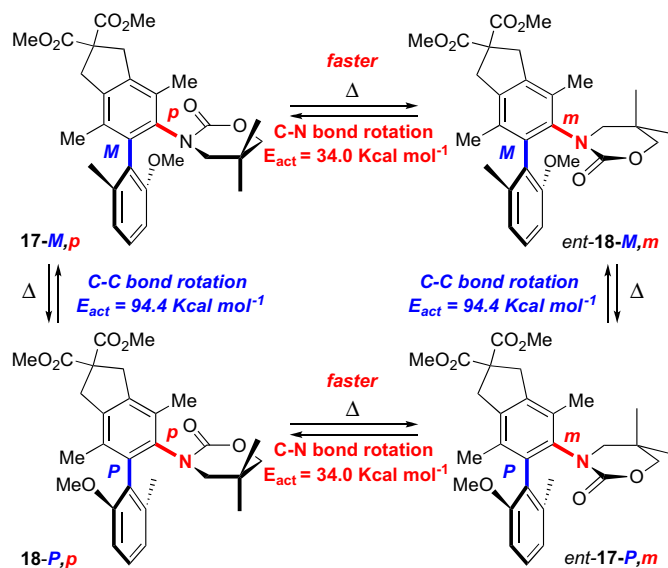
using UV and a suitable chemical stain. Low-resolution mass spectra were obtained using an Agilent-1100-HPLC/MSD and can be either APCI or ESI, or an IonSpec HiRes-MALDI FT-Mass Spectrometer. High-resolution mass spectral analyses were performed at University of Wisconsin Mass Spectrometry Laboratories. All spectral data obtained for new compounds are reported.

4.2. A general procedure for the asymmetric ynamide-[2+2+2] cycloaddition

To a solution of [Rh(cod)₂]BF₄ (10 mol%) and (*S*)-xylyl-BINAP (10 mol%) in anhyd ClCH₂CH₂Cl (5.0 mM) was added 4 Å molecular sieves in a sealed tube. The mixture was stirred at rt for 10 min before the respective ynamide (1.00 mmol) and diyne (2.00 mmol) were added. The solution was heated to 85 °C and followed by LCMS. After the reaction was complete, the solution was cooled to rt and filtered through a short pad of silica gel. Elution with EtOAc/



Scheme 7. Synthesis of chiral amino-biaryl **38-M**.



Scheme 8. Thermal equilibration and rotational barriers.

hexanes (1:1) followed by concentration in vacuo afforded a crude mixture of diastereomers. Separation and purification of the resulting crude residue via silica gel flash column chromatography (gradient eluent: EtOAc in hexanes) afforded the desired *N,O*-biaryl diastereomers. Diastereomeric ratios were found in crude ^1H NMR, and enantiomeric excess of each diastereomer was determined via chiral HPLC [CHIRALCEL OD-H; size: 250×4.6 mm (*L*×*i.d.*); eluent: isopropyl alcohol in Hexanes].

4.2.1. Establishing the feasibility

4.2.1.1. Chiral biaryl 12-major. $R_f=0.52$ [20% EtOAc in chloroform]; thick oil; ^1H NMR (500 MHz, CDCl_3) δ 1.81 (s, 3H), 1.96 (s, 3H), 2.16 (s, 3H), 3.35 (td, $J=8.9, 4.6$ Hz, 1H), 3.54–3.59 (m, 2H), 3.61–3.74 (m, 4H), 3.68 (s, 3H), 3.792 (s, 3H), 3.797 (s, 3H), 4.22 (td, $J=8.6, 4.6$ Hz, 1H), 6.77 (d, $J=8.5$ Hz, 1H), 6.90 (d, $J=8.0$ Hz, 1H), 7.23 (t, $J=8.0$ Hz, 1H); ^{13}C NMR (125 MHz, CDCl_3) δ 15.0, 16.4, 20.0, 40.66, 40.68, 46.9, 53.36, 53.43, 55.6, 59.4, 62.9, 107.7, 123.4, 126.8, 128.6, 129.9, 131.1, 133.2, 136.2, 139.3, 139.4, 139.7, 156.6, 156.8, 172.4, 172.8; IR (film) cm^{-1} 2946w, 2354w, 1730s, 1688s, 1461m, 1439m, 1235s, 1183m; mass spectrum (APCI): m/e (% relative intensity) 468 (100) ($\text{M}+\text{H}$) $^+$. **Compound 12-Minor:** $R_f=0.52$ [20% EtOAc in chloroform]; thick oil; ^1H NMR (500 MHz, CDCl_3) δ 1.88 (s, 3H), 1.93 (s, 3H), 2.17 (s, 3H), 3.01 (td, $J=8.6, 4.7$ Hz, 1H), 3.48 (q, $J=8.8$ Hz, 1H), 3.54–3.75 (m, 5H),

Table 1
Heating of pure **17-*M,p*** diastereomer

temp [°C]	time [h]	17-<i>M,p</i> [ee]	ent-17-<i>P,m</i>	ent-18-<i>M,m</i> [ee]	18-<i>P,p</i>
starting 17 [95%]					
Δ at 85	after 24	97.5 [95%]	2.5	0.0	0.0
isomerizing via C-N					
120	24	58.2 [94%]	1.8	38.8 [94%]	1.2
isomerizing via C-N					
165	24	24.0 [92%]	1.0	71.3 [90%]	3.7
200	24	22.1 [92%]	0.9	72.8 [89%]	4.2
via C-C					
final dr of 17 : ent-18 = 23 : 77					

Table 2
Heating of pure **ent-17-*P,m*** and **18-*P,p*** diastereomers

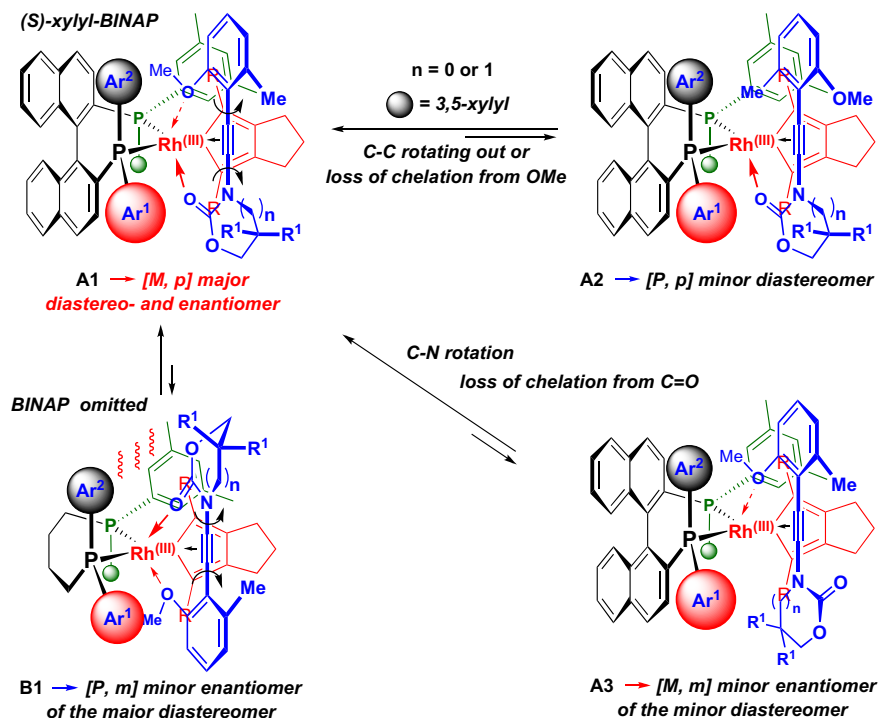
temp [°C]	time [h]	17-<i>M,p</i>	ent-17-<i>P,m</i> [ee]	ent-18-<i>M,m</i>	18-<i>P,p</i> [ee]
starting ent-17 [86%]					
Δ at 85	after 24	7.0	93.0 [86%]	0.0	0.0
C-N					
120	24	1.8	24.2 [86%]	5.2	68.8 [86%]
C-N					
free rotations at both C-N and C-C					
165	24	1.8	21.2 [84%]	7.7	69.3 [80%]
final dr of ent-17 : 18 = 23 : 77					
free rotations at both C-N and C-C					
Δ at 165	after 24	6.2	15.8 [44%]	20.3	18 [52%]
200	8	6.7	15.3 [39%]	21.8	57.7 [48%]
final dr of ent-17:18 = 22 : 78					

3.67 (m, 3H), 3.78 (s, 3H), 3.81 (s, 3H), 4.18 (td, $J=8.7, 4.7$ Hz, 1H), 6.81 (d, $J=8.4$ Hz, 1H), 6.85 (d, $J=7.2$ Hz, 1H), 7.24 (t, $J=8.0$ Hz, 1H); ^{13}C NMR (125 MHz, CDCl_3) δ 15.0, 16.4, 19.9, 30.0, 40.7, 46.3, 53.4, 55.9, 59.4, 62.3, 108.5, 122.1, 126.3, 126.6, 128.8, 130.06, 130.08, 131.1, 135.6, 139.5, 139.8, 156.4, 157.8, 172.2, 173.0 [missing one carbon due to overlap]; IR (film) cm^{-1} 2896w, 2344w, 1724s, 1679s, 1443m, 1425m, 1245s, 1190m; mass spectrum (APCI): m/e (% relative intensity) 468 (100) ($\text{M}+\text{H}$) $^+$; HRMS of the isomeric mixture (ESI, m/e) calcd for $\text{C}_{26}\text{H}_{30}\text{NO}_7$ 468.2017, found 468.2013.

4.2.1.2. Chiral biaryl 14-M. Rotamers observed on the NMR time-scale; $R_f=0.34$ [60% EtOAc in hexanes]; pale solid; mp 185–188 °C; $[\alpha]_D^{20} -16.7$ (c 2.14, CH_2Cl_2); ^1H NMR (400 MHz, CDCl_3) δ 0.53 (s, 3H), 0.73 (s, 3H), 1.25 (s, 3H), 1.26 (s, 3H), 2.07 (s, 3H), 2.17 (s, 3H), 3.66 (d, $J=8.4$ Hz, 1H), 3.68 (s, 6H), 3.75 (d, $J=8.4$ Hz, 1H), 3.97 (d, $J=8.0$ Hz, 1H), 3.99 (d, $J=8.0$ Hz, 1H), 5.09–5.20 (m, 8H), 6.73 (d, $J=8.4$ Hz, 1H), 6.77 (d, $J=8.4$ Hz, 1H), 6.85 (d, $J=6.8$ Hz, 1H), 6.87 (d, $J=7.6$ Hz, 1H), 7.13 (s, 2H), 7.15 (s, 1H), 7.19 (s, 1H), 7.22 (t, $J=8.0$ Hz, 1H), 7.23 (t, $J=8.0$ Hz, 1H); ^{13}C NMR (100 MHz, CDCl_3) δ 20.2, 21.1, 23.5, 23.8, 24.9, 55.3, 55.6, 62.2, 62.5, 73.5, 73.6, 75.0, 75.3, 107.7, 107.8, 122.69, 122.72, 124.5, 125.0, 125.2, 125.5, 127.9, 128.2, 128.7, 128.9, 133.8, 134.1, 137.4, 137.5, 138.0, 139.76, 139.83, 139.91, 139.95, 157.3, 157.6, 158.4 [missing four carbons due to overlap]; IR (film) cm^{-1} 2935w, 1746s, 1466m, 1389s, 1259s, 1068s, 1029s, 734s; mass spectrum (APCI): m/e (% relative intensity) 354 (100) ($\text{M}+\text{H}$) $^+$; HRMS (ESI, m/e) calcd for $\text{C}_{21}\text{H}_{23}\text{NO}_4$ 353.1622, found 353.1626; HPLC (80:20 hexane/2-propanol): t_R =major 15.6 min, and minor 12.0 min.

4.2.1.3. Chiral biaryl 15-M. $R_f=0.10$ [50% EtOAc in hexanes]; pale solid; mp 96–100 °C; $[\alpha]_D^{20} -27.5$ (c 2.0, CH_2Cl_2); ^1H NMR (500 MHz, CDCl_3) δ 0.58 (br s, 3H), 0.92–0.96 (br m, 3H), 2.09 (s, 3H), 3.02–3.09 (br m, 2H), 3.62–3.70 (br m, 1H), 3.70 (s, 3H), 3.84–3.87 (br m, 1H), 5.12 (s, 2H), 5.16 (s, 2H), 6.78 (d, $J=8.5$ Hz, 1H), 6.89 (d, $J=7.5$ Hz, 1H), 7.04 (s, 1H), 7.23 (t, $J=8.0$ Hz, 1H), 7.27 (s, 1H); ^{13}C NMR (125 MHz, CDCl_3) δ 20.8, 22.0, 22.8, 29.2, 31.1, 55.9, 59.9, 73.9, 76.3, 108.0, 122.0, 123.2, 124.7, 127.4, 129.5, 135.6, 139.4, 140.7, 140.9, 141.5, 152.3, 156.8; IR (film) cm^{-1} 2959w, 2927w, 2361w, 2249w, 1735s, 1698s, 1476m, 1356m; mass spectrum (APCI): m/e (% relative intensity) 368 (100) ($\text{M}+\text{H}$) $^+$.

4.2.1.4. Chiral biaryl 16-M. $R_f=0.38$ [60% EtOAc in hexanes]; pale solid; mp 74–80 °C; $[\alpha]_D^{20} -25.3$ (c 2.16, CH_2Cl_2); ^1H NMR (400 MHz, CDCl_3) δ 0.56 (br s, 3H), 0.98 (br m, 3H), 2.09 (s, 3H), 2.12 (quintet,



Scheme 9. A proposed asymmetric model.

$J=7.6$ Hz, 2H), 2.89–3.08 (br m, 6H), 3.65 (br m, 1H), 3.69 (s, 3H), 3.86–3.87 (br m, 1H), 6.76 (d, $J=8.4$ Hz, 1H), 6.87 (d, $J=7.6$ Hz, 1H), 7.01 (s, 1H), 7.20 (t, $J=8.0$ Hz, 1H), 7.22 (s, 1H); ^{13}C NMR (100 MHz, CDCl_3) δ 20.4, 21.8, 22.9, 25.8, 29.0, 32.8, 33.0, 55.6, 59.9, 76.4, 107.9, 123.1, 124.6, 127.4, 128.0, 128.5, 133.6, 139.3, 140.5, 144.2, 145.0, 152.7, 156.9; IR (film) cm^{-1} 2959w, 2839w, 1791s, 1473s, 1371m, 1260s, 1200s, 1148m, 1057m; mass spectrum (APCI): m/e (% relative intensity) 366 (100) ($\text{M}+\text{H}^+$).

4.2.2. Synthesis of chiral N,O-biaryls

4.2.2.1. Chiral biaryl 17-M,p. $R_f=0.28$ [60% EtOAc in hexanes]; pale solid; mp 111–113 °C; $[\alpha]_D^{20} -46.7$ (c 1.0, CH_2Cl_2). Compound **ent-17-P,m**: $[\alpha]_D^{20} +39.5$ (c 0.72, CH_2Cl_2); ^1H NMR (500 MHz, CDCl_3) δ 0.53 (s, 3H), 1.01 (s, 3H), 1.85 (s, 3H), 1.91 (s, 3H), 2.15 (s, 3H), 2.57 (d, $J=11.0$ Hz, 1H), 2.82 (dd, $J=11.2, 1.2$ Hz, 1H), 3.34 (d, $J=10.5$ Hz, 1H), 3.59 (ABq, $\Delta\nu=19.7$ Hz, $J=17.0$ Hz, 2H), 3.65 (ABq, $\Delta\nu=18.2$ Hz, $J=17.0$ Hz, 2H), 3.67 (s, 3H), 3.70 (dd, $J=10.5, 1.0$ Hz, 1H), 3.78 (s, 3H), 3.80 (s, 3H), 6.79 (d, $J=8.5$ Hz, 1H), 6.83 (d, $J=7.5$ Hz, 1H), 7.23 (t, $J=8.0$ Hz, 1H); ^{13}C NMR (125 MHz, CDCl_3) δ 14.9, 16.5, 20.0, 22.4, 23.2, 29.2, 40.7, 40.8, 53.3, 53.4, 55.9, 58.7, 59.4, 75.4, 108.3, 122.0, 126.7, 128.7, 129.6, 131.2, 134.7, 137.7, 137.9, 138.9, 139.2, 151.6, 158.3, 172.2, 173.1; IR (Neat) cm^{-1} 2958w, 2360m, 2342m, 1734m, 1696m, 1433w, 1265s, 1109m, 734s; mass spectrum (APCI): m/e (% relative intensity) 510 (100) ($\text{M}+\text{H}^+$); HRMS (ESI, m/e) calcd for $\text{C}_{29}\text{H}_{35}\text{NO}_7$ 510.2487, found 510.2482; HPLC (95:5 hexane/2-propanol): t_R =major 11.4 min, and minor 14.7 min.

4.2.2.2. Chiral biaryl 18-P,p. $R_f=0.35$ [60% EtOAc in hexanes]; pale solid; mp 187–190 °C; $[\alpha]_D^{20} +5.9$ (c 2.40, CH_2Cl_2). Compound **ent-18-M,m**: $[\alpha]_D^{20} -3.2$ (c 0.12, CH_2Cl_2); ^1H NMR (400 MHz, CDCl_3) δ 0.51 (s, 3H), 0.95 (s, 3H), 1.78 (s, 3H), 2.01 (s, 3H), 2.15 (s, 3H), 2.92 (d, $J=11.6$ Hz, 1H), 2.99 (dd, $J=11.8, 1.6$ Hz, 1H), 3.47 (dd, $J=10.4, 1.6$ Hz, 1H), 3.65 (ABq, $\Delta\nu=71.6$ Hz, $J=16.8$ Hz, 2H), 3.66 (ABq, $\Delta\nu=24.1$ Hz, $J=17.2$ Hz, 2H), 3.67 (s, 3H), 3.789 (s, 3H), 3.795 (s, 3H), 3.80 (d, $J=10.5$ Hz, 1H), 6.75 (d, $J=8.0$ Hz, 1H), 6.89 (d, $J=7.6$ Hz, 1H), 7.21 (t, $J=8.0$ Hz, 1H); ^{13}C NMR (125 MHz, CDCl_3) δ 14.9, 16.4, 20.2,

21.9, 23.2, 29.1, 40.7, 40.8, 53.3, 53.4, 55.3, 58.8, 59.4, 76.0, 107.7, 123.5, 126.8, 128.5, 129.2, 131.4, 135.1, 138.0, 138.7, 139.2, 140.5, 152.1, 156.7, 172.4, 172.9; IR (Neat) cm^{-1} 2957w, 2360w, 1734s, 1697s, 1467m, 1427m, 1251s, 1172s, 1061s; mass spectrum (APCI): m/e (% relative intensity) 510 (100) ($\text{M}+\text{H}^+$); HRMS (ESI, m/e) calcd for $\text{C}_{29}\text{H}_{35}\text{NO}_7$ 510.2487, found 510.2494; HPLC (95:5 hexane/2-propanol): t_R =major 40.9 min and minor 40.0 min.

4.2.2.3. Chiral biaryl 19-M,p. $R_f=0.34$ [60% EtOAc in hexanes]; pale solid; mp 80–83 °C; $[\alpha]_D^{20} -65.1$ (c 0.723, CH_2Cl_2); ^1H NMR (400 MHz, CDCl_3) δ 0.54 (s, 3H), 1.02 (s, 3H), 1.86 (s, 3H), 1.89–1.92 (m, 1H), 1.94 (s, 3H), 2.00–2.15 (m, 2H), 2.16 (s, 3H), 2.60 (d, $J=10.4$ Hz, 1H), 2.83–2.94 (m, 4H), 3.35 (d, $J=10.0$ Hz, 1H), 3.69 (s, 3H), 3.71 (dd, $J=6.8, 1.2$ Hz, 1H), 6.80 (d, $J=8.4$ Hz, 1H), 6.84 (d, $J=7.6$ Hz, 1H), 7.22 (t, $J=8.0$ Hz, 1H); ^{13}C NMR (100 MHz, CDCl_3) δ 15.0, 16.6, 20.0, 22.3, 23.2, 24.4, 29.2, 32.67, 32.70, 55.9, 58.7, 75.4, 108.3, 121.9, 127.2, 128.4, 129.3, 131.0, 133.5, 136.6, 137.9, 143.0, 143.3, 151.7, 158.3; IR (film) cm^{-1} 2949s, 2923s, 2852s, 2362m, 2344m, 1685s, 1469s, 1373m, 1256m; mass spectrum (APCI): m/e (% relative intensity) 394 (100) ($\text{M}+\text{H}^+$); HRMS of mixture of **19-M,p** and **20-P,p** (ESI, m/e) calcd for $\text{C}_{25}\text{H}_{31}\text{NO}_3$ 393.2299, found 393.2303; HPLC (95:5 hexane/2-propanol): t_R =major 26.7 min and minor 25.5 min.

4.2.2.4. Chiral biaryl 20-P,p. $R_f=0.48$ [60% EtOAc in hexanes]; thick oil; $[\alpha]_D^{20} +7.5$ (c 0.185, CH_2Cl_2); ^1H NMR (400 MHz, CDCl_3) δ 0.44 (s, 3H), 0.88 (s, 3H), 1.70–1.76 (m, 1H), 1.72 (s, 3H), 1.97 (s, 3H), 2.05 (quintet, $J=7.4$ Hz, 2H), 2.10 (s, 3H), 2.80–2.95 (m, 4H), 3.50 (dd, $J=11.6, 1.8$ Hz, 1H), 3.41 (dd, $J=10.6, 1.8$ Hz, 1H), 3.62 (s, 3H), 3.74 (d, $J=10.8$ Hz, 1H), 6.69 (d, $J=8.4$ Hz, 1H), 6.83 (d, $J=7.6$ Hz, 1H), 7.13 (t, $J=8.0$ Hz, 1H); ^{13}C NMR (100 MHz, CDCl_3) δ 15.0, 16.5, 20.2, 21.9, 23.2, 24.3, 29.0, 29.9, 32.67, 32.75, 55.4, 58.9, 76.0, 107.7, 123.4, 127.3, 128.3, 131.2, 133.8, 137.0, 140.6, 142.9, 143.5, 152.2, 156.7; IR (film) cm^{-1} 2969s, 2918s, 2850s, 2360m, 2340m, 1686s, 1467s, 1378m, 1256m; mass spectrum (APCI): m/e (% relative intensity) 394 (100) ($\text{M}+\text{H}^+$); HRMS of mixture of **19-M,p** and **20-P,p** (ESI, m/e) calcd for $\text{C}_{25}\text{H}_{31}\text{NO}_3$ 393.2299, found 393.2303; HPLC (95:5 hexane/2-propanol): t_R =major 26.7 min and minor 25.5 min.

4.2.2.5. Chiral biaryl 21-M,p. $R_f=0.17$ [60% EtOAc in hexanes]; pale solid; mp 174–176 °C; $[\alpha]_D^{20} -71.3$ (c 1.42, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 0.55 (s, 3H), 1.03 (s, 3H), 1.83 (s, 3H), 1.95 (s, 3H), 2.14 (s, 3H), 2.62 (d, $J=10.4$ Hz, 1H), 2.85 (dd, $J=11.2, 1.6$ Hz, 1H), 3.37 (d, $J=10.4$ Hz, 1H), 3.69 (s, 3H), 3.72 (dd, $J=10.6, 1.4$ Hz, 1H), 5.07–5.16 (m, 4H), 6.82 (d, $J=8.0$ Hz, 1H), 6.86 (d, $J=7.6$ Hz, 1H), 7.25 (t, $J=8.0$ Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 14.9, 16.3, 19.8, 22.3, 23.0, 29.1, 55.8, 58.6, 74.3, 75.4, 108.3, 122.0, 126.0, 127.3, 128.8, 128.9, 135.2, 137.7, 137.99, 138.02, 138.5, 151.5, 158.1 [missing one carbon due to overlap]; IR (film) cm⁻¹ 3418w, 2967w, 2360w, 1692s, 1578w, 1468s, 1373m, 1259m, 1175m; mass spectrum (APCI): m/e (% relative intensity) 396 (100) (M+H)⁺; HRMS (ESI, m/e) calcd for C₂₄H₂₉NO₄ 395.2092, found 395.2093; HPLC (95:5 hexane/2-propanol): t_R =major 27.0 min and minor 26.5 min.

4.2.2.6. Chiral biaryl 22-P,p. $R_f=0.26$ [60% EtOAc in hexanes]; thick oil; $[\alpha]_D^{20} +21.4$ (c 0.075, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 0.52 (s, 3H), 0.96 (s, 3H), 1.76 (s, 3H), 2.04 (s, 3H), 2.14 (s, 3H), 2.95 (d, $J=11.6$ Hz, 1H), 3.01 (dd, $J=11.6, 1.6$ Hz, 1H), 3.50 (dd, $J=10.8, 2.0$ Hz, 1H), 3.70 (s, 3H), 3.83 (d, $J=10.4$ Hz, 1H) 5.14–5.19 (m, 4H), 6.77 (d, $J=8.4$ Hz, 1H), 6.95 (d, $J=7.6$ Hz, 1H), 7.24 (t, $J=8.0$ Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 14.9, 16.3, 20.1, 21.8, 23.2, 29.0, 55.3, 58.7, 74.3, 76.0, 77.4, 107.8, 123.5, 126.2, 127.0, 128.7, 129.1, 135.6, 137.9, 138.4, 138.6, 140.4, 152.0, 156.6; IR (film) cm⁻¹ 2969w, 2884w, 2839w, 2360m, 2342m, 1754s, 1699s, 1468s, 1258s; mass spectrum (APCI): m/e (% relative intensity) 396 (100) (M+H)⁺; HRMS (ESI, m/e) calcd for C₂₄H₂₉NO₄ 395.2092, found 395.2081; HPLC (95:5 hexane/2-propanol): t_R =major 24.3 min and minor 25.8 min.

4.2.2.7. Chiral biaryl 23-M,p. $R_f=0.32$ [60% EtOAc in hexanes]; pale solid; mp 96–100 °C; $[\alpha]_D^{20} -6.8$ (c 1.18, CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ 0.45 (s, 3H), 1.14 (s, 3H), 1.88 (s, 3H), 2.00 (s, 3H), 2.14 (s, 3H), 3.61 (s, 2H), 3.65 (ABq, $\Delta\nu=17.1$ Hz, $J=16.8$ Hz, 2H), 3.66 (s, 3H), 3.75 (d, $J=8.0$ Hz, 1H), 3.79 (s, 3H), 3.80 (s, 3H), 4.00 (d, $J=8.0$ Hz, 1H), 6.77 (d, $J=8.0$ Hz, 1H), 6.81 (d, $J=7.5$ Hz, 1H), 7.21 (d, $J=8.0$ Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 16.9, 17.5, 20.2, 24.7, 25.8, 40.9, 41.2, 53.38, 53.40, 55.8, 59.2, 62.0, 77.0, 108.2, 122.1, 127.7, 128.8, 131.79, 131.85, 133.1, 137.0, 139.3, 139.7, 139.8, 157.2, 158.0, 172.2, 173.1; IR (film) cm⁻¹ 2972w, 2360w, 1735s, 1434m, 1374m, 1264s, 1061m, 733s; mass spectrum (APCI): m/e (% relative intensity) 496 (100) (M+H)⁺; HRMS (ESI, m/e) calcd for C₂₈H₃₃NO₇ 496.2330, found 496.2339; HPLC (80:20 hexane/2-propanol): t_R =major 28.2 min and minor 25.8 min.

4.2.2.8. Chiral biaryl 24-P,p. $R_f=0.44$ [60% EtOAc in hexanes]; thick oil; $[\alpha]_D^{20} +29.8$ (c 0.33, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 0.61 (s, 3H), 1.10 (s, 3H), 1.77 (s, 3H), 2.01 (s, 3H), 2.12 (s, 3H), 3.65 (ABq, $\Delta\nu=61.3$, $J=16.8$ Hz, 2H), 3.67 (ABq, $\Delta\nu=31.7$ Hz, $J=17.2$ Hz, 2H), 3.69 (s, 3H), 3.796 (s, 3H), 3.804 (s, 3H), 3.89 (d, $J=8.0$ Hz, 1H), 4.05 (d, $J=8.0$ Hz, 1H), 6.69 (d, $J=8.4$ Hz, 1H), 6.87 (d, $J=7.6$ Hz, 1H), 7.20 (d, $J=8.0$ Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 16.8, 17.0, 20.1, 23.1, 26.1, 40.9, 41.1, 53.3, 53.4, 55.1, 59.1, 61.9, 107.3, 123.2, 127.9, 128.6, 131.1, 131.7, 133.2, 137.0, 139.2, 139.5, 139.8, 157.8, 157.9, 172.4, 172.9, [missing one carbon due to overlap]; IR (film) cm⁻¹ 2956w, 2360w, 1735s, 1698s, 1468m, 1433m, 1252s, 1173m; mass spectrum (APCI): m/e (% relative intensity) 496 (100) (M+H)⁺; HRMS (ESI, m/e) calcd for C₂₈H₃₃NO₇ 496.2330, found 496.2320; HPLC (95:5 hexane/2-propanol): t_R =major 25.5 min and minor 19.3 min.

4.2.2.9. Chiral biaryl 25-M,p. $R_f=0.35$ [60% EtOAc in hexanes]; pale solid; mp 110–113 °C; $[\alpha]_D^{20} -4.3$ (c 3.15, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 0.48 (s, 3H), 1.17 (s, 3H), 1.86 (s, 3H), 2.04 (s, 3H), 2.13 (s, 3H), 3.68 (s, 3H), 3.79 (d, $J=8.4$ Hz, 1H), 4.03 (d, $J=8.0$ Hz, 1H), 5.07–5.19 (m, 4H), 6.79 (d, $J=8.4$ Hz, 1H), 6.83 (d, $J=7.6$ Hz, 1H), 7.24 (t, $J=8.0$ Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 16.7, 17.2, 20.1,

24.6, 25.8, 55.8, 61.9, 74.5, 74.6, 108.3, 122.2, 127.0, 129.0, 129.5, 129.6, 133.6, 137.5, 139.0, 139.06, 139.1, 157.1, 157.9 [missing one carbon due to overlap]; IR (film) cm⁻¹ 2968w, 2858w, 2369w, 2335w, 1745s, 1582w, 1397m, 1210m; mass spectrum (APCI): m/e (% relative intensity) 382 (100) (M+H)⁺; HRMS (ESI, m/e) calcd for C₂₃H₂₇NO₄ 381.1935, found 381.1938; HPLC (95:5 hexane/2-propanol): t_R =major 27.0 min and minor 26.0 min.

4.2.2.10. Chiral biaryl 26-P,p. $R_f=0.41$ [60% EtOAc in hexanes]; thick oil; $[\alpha]_D^{20} +65.0$ (c 0.85, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 0.63 (s, 3H), 1.13 (s, 3H), 1.76 (s, 3H), 2.05 (s, 3H), 2.11 (s, 3H), 3.72 (s, 3H), 3.92 (d, $J=8.4$ Hz, 1H), 4.07 (d, $J=8.0$ Hz, 1H), 5.12–5.21 (m, 4H), 6.72 (d, $J=8.4$ Hz, 1H), 6.90 (d, $J=7.6$ Hz, 1H), 7.23 (t, $J=8.0$ Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 16.6, 16.9, 20.1, 23.1, 26.1, 55.2, 62.0, 74.5, 74.6, 77.7, 107.4, 123.3, 127.3, 128.8, 128.9, 129.6, 133.7, 137.6, 138.5, 139.0, 139.7, 157.80, 157.85; IR (film) cm⁻¹ 2956w, 2360w, 1735s, 1698s, 1468m, 1433m, 1252s, 1173m; mass spectrum (APCI): m/e (% relative intensity) 496 (100) (M+H)⁺; HRMS (ESI, m/e) calcd for C₂₃H₂₇NO₄ 382.2013, found 382.2013; HPLC (95:5 hexane/2-propanol): t_R =major 24.3 min and minor 25.8 min.

4.2.2.11. Chiral biaryl 27-M,p. $R_f=0.17$ [10% EtOAc in CH₂Cl₂]; yellow solid; mp 110–114 °C; $[\alpha]_D^{20} -24.4$ (c 1.53, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ -0.096 (s, 3H), 0.99 (s, 3H), 1.73 (s, 3H), 2.18 (s, 3H), 3.65 (d, $J=7.8$ Hz, 1H), 3.66 (s, 2H), 3.71 (ABq, $\Delta\nu=19.9$ Hz, $J=17.0$ Hz, 2H), 3.81 (s, 3H), 3.82 (s, 6H), 3.94 (d, $J=8.0$ Hz, 1H), 7.23 (dd, $J=8.4, 0.8$ Hz, 1H), 7.27–7.35 (m, 2H), 7.35 (d, $J=9.2$ Hz, 1H), 7.79 (dd, $J=7.9, 1.4$ Hz, 1H), 7.87 (d, $J=9.0$ Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 16.9, 18.0, 24.1, 26.4, 40.9, 41.3, 53.4, 56.4, 59.1, 61.9, 77.0, 77.5, 113.3, 121.3, 123.3, 125.1, 127.0, 128.6, 128.7, 129.8, 131.9, 132.6, 133.9, 134.9, 135.8, 139.8, 140.1, 154.9, 157.3, 172.2, 173.1; IR (film) cm⁻¹: 2930w, 1735s, 1623w, 1595w, 1436m, 1397w, 1371m, 1351m, 1337m, 1272s, 1249s, 1167s, 1060s; mass spectrum (APCI): m/e (% relative intensity) 532 (100) (M+H)⁺, 474 (7); HRMS (ESI, m/e) calcd for C₃₁H₃₃NO₇ 532.2330, found 532.2320; HPLC (95:5 hexane/2-propanol): t_R =major 26.4 min and minor 25.2 min.

4.2.2.12. Chiral biaryl 28-P,p. $R_f=0.39$ [10% EtOAc in CH₂Cl₂]; yellow solid; mp 245–250 °C; $[\alpha]_D^{20} -19.6$ (c 1.13, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 0.38 (s, 3H), 1.12 (s, 3H), 1.80 (s, 3H), 2.17 (s, 3H), 3.50 (d, $J=8.2$ Hz, 1H), 3.60–3.80 (m [two ABq], 4H), 3.81 (s, 3H), 3.82 (s, 3H), 3.87 (s, 3H), 3.89 (d, $J=8.0$ Hz, 1H), 7.26 (d, $J=9.2$ Hz, 1H), 7.28–7.35 (m, 2H), 7.50–7.52 (m, 1H), 7.73–7.75 (m, 1H), 7.88 (d, $J=9.0$ Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 16.9, 17.2, 23.9, 25.7, 41.0, 41.2, 53.3, 53.4, 55.7, 59.1, 62.0, 77.7, 111.8, 122.2, 124.3, 126.6, 126.9, 127.2, 129.0, 129.9, 131.6, 132.7, 133.3, 133.6, 135.8, 139.5, 139.9, 154.5, 157.2, 172.4, 172.9; IR (film) cm⁻¹ 3003w, 1730s, 1592w, 1512w, 1462w, 1433w, 1397w, 1368w, 1268s, 1246s, 1199m, 1168m, 1062s; mass spectrum (APCI): m/e (% relative intensity) 532 (100) (M+H)⁺; HRMS (ESI, m/e) calcd for C₃₁H₃₃NO₇ 532.2330, found 532.2327; HPLC (95:5 hexane/2-propanol): t_R =major 20.7 min and minor 15.7 min.

4.2.2.13. Chiral biaryl 29-M,p. $R_f=0.25$ [60% EtOAc in hexanes]; yellow solid; mp 105–110 °C; $[\alpha]_D^{20} -40.8$ (c 2.055, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ -0.052 (s, 3H), 1.02 (s, 3H), 1.71 (s, 3H), 2.17 (s, 3H), 3.70 (d, $J=8.0$ Hz, 1H), 3.85 (s, 3H), 3.97 (d, $J=8.0$ Hz, 1H), 5.12–5.25 (m, 4H), 7.26–7.38 (m, 3H), 7.37 (d, $J=9.2$ Hz, 1H), 7.82 (dd, $J=7.8, 1.2$ Hz, 1H), 7.90 (d, $J=9.0$ Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 16.7, 17.9, 24.1, 26.4, 56.5, 61.9, 74.5, 74.6, 77.1, 113.3, 120.7, 123.4, 124.9, 127.2, 128.69, 128.73, 129.6, 130.1, 130.5, 134.4, 134.8, 136.4, 139.1, 139.5, 155.0, 157.3; IR (film) cm⁻¹ 2933w, 2839w, 1747s, 1620w, 1593w, 1507w, 1457w, 1416w, 1367m, 1345m, 1296w, 1274m, 1261m; mass spectrum (APCI): m/e (% relative intensity) 418 (100) (M+H)⁺; HRMS of mixture of **29-M,p** and **30-P,p** (ESI,

m/e) calcd for C₂₆H₂₇NO₄ 417.1935, found 417.1928; HPLC (80:20 hexane/2-propanol): *t_R*=major 17.1 min and minor 13.2 min.

4.2.2.14. *Chiral biaryl 30-P.p.* *R_f*=0.42 [60% EtOAc in hexanes]; yellow solid; mp 115–120 °C; [α]_D²⁰ –57.8 (c 0.79, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 0.38 (s, 3H), 1.15 (s, 3H), 1.78 (s, 3H), 2.16 (s, 3H), 3.53 (d, *J*=8.2 Hz, 1H), 3.89 (s, 3H), 3.91 (d, *J*=8.2 Hz, 1H), 5.15–5.25 (m, 4H), 7.28 (d, *J*=9.2 Hz, 1H), 7.30–7.38 (m, 2H), 7.52–7.55 (m, 1H), 7.75–7.78 (m, 1H), 7.91 (d, *J*=9.0 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 16.8, 17.1, 23.9, 25.7, 55.8, 62.0, 74.57, 74.64, 77.0, 111.8, 121.5, 124.4, 126.3, 127.1, 127.4, 129.0, 129.4, 130.2, 130.6, 133.2, 134.1, 136.4, 138.7, 139.4, 154.5, 157.2; IR (film) cm⁻¹ 2840w, 1743m, 1592w, 1510w, 1462w 1362w, 1345w, 1300w, 1258m, 1244m; mass spectrum (APCI): *m/e* (% relative intensity) 418 (100) (M+H)⁺; HRMS of mixture of **29-M.p** and **30-P.p** (ESI, *m/e*) calcd for C₂₆H₂₇NO₄ 417.1935, found 417.1928; HPLC (80:20 hexane/2-propanol): *t_R*=major 18.7 min and minor 15.2 min.

4.2.3. Assignment of absolute configurations

4.2.3.1. *Demethylation procedure.* A solution of **17-M.p** (30.0 mg, 0.059 mmol) in anhyd CH₂Cl₂ was cooled to –78 °C under an argon atmosphere. To this reaction mixture was added BBr₃ (1.0 M soln in CH₂Cl₂, 5.0 equiv) carefully dropwise. The reaction mixture was warmed to –25 °C and stirred for 24 h. After the reaction was complete, the solution was quenched with H₂O and the reaction mixture was warmed to rt. The layers were separated and the organic layer extracted with CH₂Cl₂ (2×equal volume). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated under reduced pressure.

The crude product was re-dissolved in anhyd acetone and 2.5 equiv of Me₂SO₄, and anhyd solid NaHCO₃ (2.5 equiv) was added under an argon atmosphere. The reaction mixture was stirred at rt for 48 h. After the reaction was complete, the mixture filtered through a short pad of Celite™. Acetone was removed under reduced pressure and the crude mixture was re-dissolved in EtOAc. The reaction mixture was washed with water followed by satd aq NaCl (equal volume). The organic layer was dried over MgSO₄, filtered, and concentrated under reduced pressure. Purification of the crude residue by silica gel flash column chromatography [isocratic eluent: 50% EtOAc in hexanes] yielded the desired phenol **31-M.p** in 73% yield. Compound **31-M.p**: *R_f*=0.52 [50% EtOAc in hexanes]; white solid; mp 165–168 °C; [α]_D²⁰ +7.8 (c 2.05, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 0.50 (s, 3H), 0.96 (s, 3H), 1.84 (s, 3H), 1.94 (s, 3H), 2.16 (s, 3H), 2.69 (dd, *J*=11.2, 2.0 Hz, 1H), 3.08 (d, *J*=11.2 Hz, 1H), 3.57–3.71 (m, 5H), 3.79 (s, 3H), 3.80 (s, 3H), 3.94 (d, *J*=10.8 Hz, 1H), 6.28 (s, 1H), 6.82 (d, *J*=7.2 Hz, 1H), 6.89 (d, *J*=7.6 Hz, 1H), 7.14 (t, *J*=8.0 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 14.5, 16.4, 20.2, 21.3, 23.0, 29.1, 40.6, 53.3, 53.4, 58.4, 59.3, 60.6, 76.1, 116.2, 122.5, 125.8, 129.0, 129.2, 132.5, 133.5, 136.7, 137.8, 139.9, 140.3, 154.1, 155.0, 172.0, 172.7 [missing one carbon due to overlap]; IR (film) cm⁻¹ 3332s, 2959w, 2889w, 2360w, 2254w, 1734s, 1697s, 1467m, 1402m; mass spectrum (APCI): *m/e* (% relative intensity) 496 (100) (M+H)⁺; HRMS (ESI, *m/e*) calcd for C₂₈H₃₃NO₇ 496.2330, found 496.2318.

4.2.3.2. *Preparation of camphor-sulfonyl derivative.* A solution of **31-M.p** (55.0 mg, 0.11 mmol) in anhyd CH₂Cl₂ was cooled to 0 °C under an argon atmosphere. To the reaction mixture was added (1*S*)-(–)-10-camphor sulfonyl chloride (31.0 mg, 0.122 mmol) followed by catalytic DMAP and Et₃N (0.122 mmol). The reaction mixture was warmed to rt and stirred for 12 h. After the reaction was complete, the solution was quenched with pH 7.0 aqueous buffer and the layers were separated. The organic layer was extracted with CH₂Cl₂ (2×equal volume), dried over Na₂SO₄, filtered, and concentrated under reduced pressure. Purification of the crude residue by silica gel flash column chromatography

[isocratic eluent: 40% EtOAc in hexanes] afforded the product **32-M.p** in 85% yield. Compound **32-M.p**: *R_f*=0.46 [50% EtOAc in hexanes]; clear crystals; mp 230–233 °C; [α]_D²⁰ –39.2 (c 1.66, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 0.49 (s, 3H), 0.73 (s, 3H), 0.79 (s, 3H), 0.97 (s, 3H), 1.20–1.28 (m, 1H), 1.36–1.43 (m, 1H), 1.52–1.59 (m, 1H), 1.71–1.80 (m, 1H), 1.80 (d, *J*=18.4 Hz, 1H), 1.89 (s, 3H), 1.97 (t, *J*=4.4 Hz, 1H), 2.01 (s, 3H), 2.15 (s, 3H), 2.27 (ddd, *J*=17.2, 4.4, 3.2 Hz, 1H), 2.47 (d, *J*=10.8 Hz, 1H), 2.84 (d, *J*=10.8 Hz, 1H), 3.40 (ABq, $\Delta\nu$ =85.5 Hz, *J*=15.2 Hz, 2H), 3.49 (d, *J*=10.0 Hz, 1H), 3.58 (ABq, $\Delta\nu$ =33.3 Hz, *J*=17.0 Hz, 2H), 3.66 (ABq, $\Delta\nu$ =54.1 Hz, *J*=16.6 Hz, 2H), 3.78 (d, *J*=9.2 Hz, 1H), 3.79 (s, 3H), 3.80 (s, 3H), 7.17 (d, *J*=7.6 Hz, 1H), 7.29 (t, *J*=8.0 Hz, 1H), 7.44 (d, *J*=8.4 Hz, 1H); ¹³C NMR (400 MHz, CDCl₃) δ 15.1, 16.7, 19.4, 19.6, 20.0, 22.1, 23.1, 24.0, 27.1, 29.1, 40.7, 42.6, 43.1, 48.3, 49.7, 53.3, 53.4, 58.3, 58.8, 59.0, 75.6, 120.1, 127.9, 129.0, 130.1, 130.8, 132.0, 132.8, 138.3, 138.6, 139.1, 139.8, 147.4, 151.4, 172.2, 172.7, 214.1; IR (film) cm⁻¹ 2959w, 2889w, 2360w, 2254w, 1734s, 1697s, 1467m, 1402m; mass spectrum (APCI): *m/e* (% relative intensity) 709 (100) (M+H)⁺; HRMS (ESI, *m/e*) calcd for C₃₈H₄₇NO₁₀S 710.2994, found 710.3004.

4.2.3.3. *Free phenol 33-M.p.* ¹H NMR (400 MHz, CDCl₃) δ 0.66 (s, 3H), 1.17 (s, 3H), 1.87 (s, 3H), 1.99 (s, 3H), 2.14 (s, 3H), 3.59–3.72 (m, 3H), 3.81 (s, 3H), 3.81 (s, 3H), 6.05 (s, 1H), 6.79 (d, 1H, *J*=7.6 Hz), 6.89 (d, *J*=8.0 Hz), 7.14 (t, *J*=8.0 Hz, 1H); mass spectrum (APCI): *m/e* (% relative intensity) 496 (10) (M+H)⁺, 483 (30), 482 (100). Compound **34-M.p**: ¹H NMR (500 MHz, CDCl₃) δ 0.47 (s, 3H), 0.75 (s, 3H), 0.84 (s, 3H), 1.11 (s, 3H), 1.40 (ddd, *J*=4.4, 9.2, 13.6 Hz, 1H), 1.65–1.83 (m, 4H), 1.92 (s, 3H), 1.97 (t, *J*=4.4 Hz, 1H), 2.11 (s, 3H), 2.15 (s, 3H), 2.30 (ddd, *J*=2.8, 4.4, 18.4 Hz, 1H), 2.93 (d, *J*=15.2 Hz, 1H), 3.61 (t, *J*=6.4 Hz, 2H), 3.66 (d, *J*=6.8 Hz, 2H), 3.80 (s, 6H), 4.03 (d, *J*=8.0 Hz, 1H), 7.15 (d, *J*=7.2 Hz, 1H), 7.28 (t, *J*=9.2 Hz, 1H), 7.36 (d, *J*=7.6 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 17.0, 17.5, 19.5, 19.6, 20.3, 24.4, 25.9, 27.1, 40.9, 41.1, 42.6, 43.3, 48.0, 49.6, 53.3, 53.4, 58.4, 58.8, 62.1, 120.8, 128.1, 129.0, 132.2, 132.3, 132.5, 133.5, 135.3, 137.2, 139.5, 140.3, 140.6, 147.3, 153.9, 157.2, 169.4, 172.2, 172.8; IR (neat) cm⁻¹ 3474w, 2967m, 2893m, 23841w, 2340w, 1737s, 1686m, 1578w, 1481m, 1433s, 1398m, 1374m, 1355m; mass spectrum (APCI): *m/e* (% relative intensity) 695 (40) (M+H)⁺, 694 (100), 481 (20), 480 (90), 422 (20), 231(20); HRMS (ESI, *m/e*) calcd for C₃₇H₄₅NNaO₁₀S 718.2657, found 718.2643.

4.2.4. Synthesis of a chiral amino-biaryl ligand

4.2.4.1. *Ynamide 35.* *R_f*=0.58 [50% EtOAc in hexanes]; ¹H NMR (400 MHz, CDCl₃) δ 2.12 (s, 3H), 3.72 (s, 3H), 4.88 (s, 2H), 6.62 (d, *J*=8.0 Hz, 1H), 6.70 (d, *J*=7.6 Hz, 1H), 7.07 (t, *J*=8.0 Hz, 1H), 7.36–7.41 (m, 6H), 7.41–7.46 (m, 4H); ¹³C NMR (100 MHz, CDCl₃) δ 20.9, 53.7, 56.0, 70.4, 71.8, 76.6, 86.0, 108.0, 111.9, 121.8, 127.7, 128.7, 128.9, 139.7, 142.1, 155.3, 160.5 [missing eight carbons due to symmetry and overlap]; IR (film) cm⁻¹ 2969w, 2900w, 2839w, 2360w, 2248m, 1777w, 1716s, 1472s; mass spectrum (APCI): *m/e* (% relative intensity) 384 (100) (M+H)⁺; HRMS (ESI, *m/e*) calcd for C₃₈H₄₇NO₁₀S 383.1516, found 383.1505.

4.2.4.2. *Chiral biaryl 37-M.p.* *R_f*=0.39 [60% EtOAc in hexanes]; pale solid; mp 103–106 °C; [α]_D²⁰ –21.6 (c 0.98, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 1.41 (s, 3H), 1.42 (s, 3H), 1.64 (s, 3H), 3.78 (s, 3H), 4.40 (d, *J*=8.4 Hz, 1H), 4.95 (ABqd, $\Delta\nu$ =29.9 Hz, *J*=12.4, 2.2 Hz, 2H), 5.10 (ABqd, $\Delta\nu$ =37.2 Hz, *J*=12.8, 1.6 Hz, 2H), 5.20 (d, *J*=8.4 Hz, 1H), 6.19 (d, *J*=7.6 Hz, 1H), 6.51 (d, *J*=7.6 Hz, 2H), 6.78 (d, *J*=8.0 Hz, 1H), 6.84 (t, *J*=8.0 Hz, 2H), 7.01 (d, *J*=8.0 Hz, 3H), 7.05 (t, *J*=8.0 Hz, 1H), 7.23–7.27 (m, 2H), 7.35 (t, *J*=7.6 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 15.8, 17.03, 20.0, 55.9, 71.2, 74.4, 74.6, 77.0, 107.8, 123.1, 126.5, 126.7, 127.1, 128.07, 128.15, 128.4, 128.6, 128.7, 129.8, 130.9, 135.9, 137.4, 137.9, 138.6, 138.7, 139.2, 144.0, 157.7, 158.0 [missing four carbons due to symmetry]; IR (film) cm⁻¹ 3058w, 2840w, 2360w,

1760s, 1579w, 1469m, 1262m, 1069m, 908w; mass spectrum (APCI): m/e (% relative intensity) 506 (100) (M+H)⁺; HRMS (ESI, m/e) calcd for C₃₃H₃₁NO₄ 505.2253, found 505.2250; HPLC (95:5 hexane/2-propanol): t_{R} =major 27.5 min and minor 26.4 min.

4.2.4.3. Chiral biaryl 38-P.p. R_f =0.62 [60% EtOAc in hexanes]; thick oil; $[\alpha]_D^{20}$ +74.5 (c 0.98, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 1.43 (s, 3H), 1.62 (s, 3H), 2.19 (s, 3H), 2.92 (s, 3H), 4.31 (d, J =8.0 Hz, 1H), 4.92 (ABq, $\Delta\nu$ =49.9 Hz, J =12.2 Hz, 2H), 5.12 (ABq, $\Delta\nu$ =28.7 Hz, J =12.5 Hz, 2H), 5.29 (d, J =3.2 Hz, 1H), 5.31 (d, J =4.0 Hz, 1H), 5.89 (d, J =8.4 Hz, 1H), 6.43 (br d, J =7.2 Hz, 2H), 6.86 (t, J =7.6 Hz, 1H), 6.88 (d, J =6.8 Hz, 1H), 7.01 (dt, J =6.8, 0.8 Hz, 1H), 7.04 (d, J =8.0 Hz, 1H), 7.34–7.35 (m, 4H); ¹³C NMR (100 MHz, CDCl₃) δ 15.5, 16.8, 20.2, 53.9, 71.0, 74.2, 74.5, 77.1, 107.7, 122.7, 125.4, 126.8, 127.0, 127.9, 128.2, 128.5, 128.7, 129.5, 129.9, 135.3, 136.4, 138.2, 138.9, 139.0, 139.2, 145.2, 156.7, 158.9 [missing four carbons due to symmetry and one carbon due to overlap]; IR (film) cm⁻¹ 3058w, 2839w, 2360w, 2341w, 1753s, 1598w, 1492m, 1468m, 1301m; mass spectrum (APCI): m/e (% relative intensity) 506 (100) (M+H)⁺; HRMS (ESI, m/z) calcd for C₃₃H₃₁NO₄ 505.2253, found 505.2257; HPLC (95:5 hexane/2-propanol): t_{R} =major 11.5 min and minor 16.3 min.

4.2.4.4. Preparation of chiral amino-biaryl 39-M. To a solution of chiral biaryl **37-M.p** (13.0 mg, 82% ee) in MeOH (3 mL) flushed with argon was added 10 mol% Pd/C (3.0 mg). The reaction mixture was flushed with H₂ gas and then stirred under a balloon of H₂ for 3 days. The resulting mixture was filtered through a short pad of Celite™ and solvent was removed under reduced pressure. The crude product was purified by preparative TLC (1:1 Hex/EtOAc) and 3.0 mg of the desired amino-biaryl **39-M** was isolated (42% yield). **Compound 39-M:** R_f =0.55 [10% CH₂Cl₂/EtOAc]; $[\alpha]_D^{23}$ +25.8 (c 0.0039, CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ 1.75 (s, 3H), 1.98 (s, 3H), 2.05 (s, 3H), 3.21–3.56 (br s, 1H), 3.66 (d, 1H, J =3.0 Hz), 3.73 (s, 3H), 5.15 (d, 4H, J =21.5 Hz), 6.87 (d, 1H, J =10.5 Hz), 6.96 (d, J =9.5 Hz), 7.25–7.29 (m, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 14.2, 16.5, 19.7, 29.9, 56.1, 74.4, 74.4, 108.8, 112.7, 114.1, 122.5, 123.1, 125.9, 127.5, 127.8, 128.8, 137.6, 139.3, 141.5, 141.7, 157.7; IR (neat) cm⁻¹ 2918s, 2849s, 2361m, 2340w, 1732w, 1618w, 1578w, 1467s, 1367m; mass spectrum (APCI): m/e (% relative intensity) 352 (100) (M+H)⁺, 285 (20), 284 (100).

Acknowledgements

Authors thank NIH-NIGMS [GM066055]. Authors also thank Dr. Victor Young and Mr. Ben Kucera [University of Minnesota] for providing X-ray structural analysis. Finally, we thank Professor Marisa Kozlowski for invaluable discussions.

References and notes

- For reviews on ynamides, see: (a) Zifcsak, C. A.; Mulder, J. A.; Hsung, R. P.; Rameshkumar, C.; Wei, L.-L. *Tetrahedron* **2001**, *57*, 7575; (b) Mulder, J. A.; Kurtz, K. C. M.; Hsung, R. P. *Synlett* **2003**, 1379; (c) Katritzky, A. R.; Jiang, R.; Singh, S. K. *Heterocycles* **2004**, *63*, 1455.
- For a special issue dedicated to the chemistry of ynamides, see: Chemistry of Electron-Deficient Ynamines and Ynamides. *Tetrahedron* **2006**, *62* (Tetrahedron-Symposium-in-Print).
- For chemistry of ynamides in the last two years, see: (a) Couty, S.; Liegault, B.; Meyer, C.; Cossy, J. *Tetrahedron* **2009**, *65*, 3882; (b) Deweerdt, K.; Birkedal, H.; Ruhland, T.; Skrydstrup, T. *Org. Lett.* **2009**, *11*, 221; (c) Dooleweerd, K.; Birkedal, H.; Ruhland, T.; Skrydstrup, T. *J. Org. Chem.* **2008**, *73*, 9447; (d) Saito, N.; Katayama, T.; Sato, Y. *Org. Lett.* **2008**, *10*, 3829; (e) Yasui, H.; Yorimitsu, H.; Oshima, K. *Bull. Chem. Soc. Jpn.* **2008**, *81*, 373; (f) Yasui, H.; Yorimitsu, H.; Oshima, K. *Chem. Lett.* **2008**, *37*, 40; (g) Istrate, F. M.; Buzas, A. K.; Jurberg, I. D.; Odabachian, Y.; Gagosz, F. *Org. Lett.* **2008**, *10*, 925; (h) Martínez-Esperón, M. F.; Rodríguez, D.; Castedo, L.; Saá, C. *Tetrahedron* **2008**, *64*, 3674; (i) Kim, J. Y.; Kim, S. H.; Chang, S. *Tetrahedron Lett.* **2008**, *49*, 1745; (j) Yavari, I.; Sabbaghan, M.; Hosseini, N.; Hossaini, Z. *Synlett* **2007**, 3172; (k) Hashimi, A. S. K.; Salathe, R.; Frey, W. *Synlett* **2007**, 1763; (l) Rodríguez, D.; Martínez-Esperón, M. F.; Castedo, L.; Saá, C. *Synlett* **2007**, 1963; (m) Couty, S.; Meyer, C.; Cossy, J. *Synlett* **2007**, 2819; (n) Movassaghi, M.; Hill, M. D.; Ahmad, O. K. *J. Am. Chem. Soc.* **2007**, *129*, 10096.
- For our own efforts, see: (a) Zhang, Y.; DeKorver, K. A.; Lohse, A. G.; Zhang, Y.-S.; Hsung, R. P. *Org. Lett.* **2009**, *11*, 899; (b) Al-Rashid, Z. F.; Johnson, W. L.; Hsung, R. P.; Wei, Y.; Yao, P.-Y.; Liu, R.; Zhao, K. *J. Org. Chem.* **2008**, *73*, 8780; (c) Yao, P.-Y.; Zhang, Y.; Hsung, R. P.; Zhao, K. *Org. Lett.* **2008**, *10*, 4275; (d) Zhang, X.; Hsung, R. P.; Li, H.; Zhang, Y.; Johnson, W. L.; Figueroa, R. *Org. Lett.* **2008**, *10*, 3477; (e) Al-Rashid, Z. F.; Hsung, R. P. *Org. Lett.* **2008**, *10*, 661; (f) You, L.; Al-Rashid, Z. F.; Figueroa, R.; Ghosh, S. K.; Li, G.; Lu, T.; Hsung, R. P. *Synlett* **2007**, 1656; (g) Li, H.; You, L.; Zhang, X.; Johnson, W. L.; Figueroa, R.; Hsung, R. P. *Heterocycles* **2007**, *74*, 553; (h) Zhang, X.; Hsung, R. P.; Li, H. *Chem. Commun.* **2007**, 2420; (i) Oppenheimer, J.; Johnson, W. L.; Tracey, M. R.; Hsung, R. P.; Yao, P.-Y.; Liu, R.; Zhao, K. *Org. Lett.* **2007**, *9*, 2361.
- Couty, S.; Liegault, B.; Meyer, C.; Cossy, J. *Tetrahedron* **2006**, *62*, 3882.
- Zhang, Y.; Hsung, R. P.; Zhang, X.; Huang, J.; Slafer, B. W.; Davis, A. *Org. Lett.* **2005**, *7*, 1047.
- (a) Witulski, B.; Stengel, T. *Angew. Chem., Int. Ed.* **1999**, *38*, 2426; (b) Witulski, B.; Stengel, T.; Fernández-Hernández, J. M. *Chem. Commun.* **2000**, 1965; (c) Witulski, B.; Alayrac, C. *Angew. Chem., Int. Ed.* **2002**, *41*, 3281.
- For a review, see: Zhang, Y.; Hsung, R. P. *ChemTracts* **2004**, *17*, 442.
- (a) Rainier, J. D.; Imbriglio, J. E. *J. Org. Chem.* **2000**, *65*, 7275; (b) Rainier, J. D.; Imbriglio, J. E. *Org. Lett.* **1999**, *1*, 2037.
- Tracey, M. R.; Oppenheimer, J.; Hsung, R. P. *J. Org. Chem.* **2006**, *71*, 8629.
- For a preliminary communication for this work, see: Oppenheimer, J.; Hsung, R. P.; Figueroa, R.; Johnson, W. L. *Org. Lett.* **2007**, *9*, 3969.
- For recent reviews, see: (a) Wallace, T. W. *Org. Biomol. Chem.* **2006**, *4*, 3197; (b) Bringmann, G.; Mortimer, A. J. P.; Keller, P. A.; Gresser, M. J.; Graner, J.; Breuning, M. *Angew. Chem., Int. Ed.* **2005**, *44*, 5384; (c) Varela, J. A.; Saá, C. *Chem. Rev.* **2003**, *103*, 3787; (d) Rubin, M.; Sromek, A. W.; Gervorgyan, V. *Synlett* **2003**, 2265; (e) Aubert, C.; Buisine, O.; Malacria, M. *Chem. Rev.* **2002**, *102*, 813; (f) Saito, S.; Yamamoto, Y. *Chem. Rev.* **2000**, *100*, 2901; (g) Frühauf, H. W. *Chem. Rev.* **1997**, *97*, 523; (h) Ojima, I.; Tzamarioudaki, M.; Li, Z.; Donovan, R. J. *Chem. Rev.* **1996**, *96*, 635; (i) Lautens, M.; Klute, W.; Tam, W. *Chem. Rev.* **1996**, *96*, 49.
- (a) Nishida, G.; Noguchi, K.; Hirano, M.; Tanaka, K. *Angew. Chem., Int. Ed.* **2007**, *46*, 3951; (b) Yamamoto, Y.; Ishii, J.-I.; Nishiyama, H.; Itoh, K. *J. Am. Chem. Soc.* **2005**, *127*, 9625; (c) Sato, Y.; Tamura, T.; Mori, M. *Angew. Chem., Int. Ed.* **2004**, *43*, 2436; (d) Gandon, V.; Leca, D.; Aechtner, T.; Vollhardt, K. P. C.; Malacria, M.; Aubert, C. *Org. Lett.* **2004**, *6*, 3405; (e) Kinoshita, H.; Shinokubo, H.; Oshima, K. *J. Am. Chem. Soc.* **2003**, *125*, 7784; (f) Tanaka, K.; Shirasaka, K. *Org. Lett.* **2003**, *5*, 4697; (g) Bonaga, L. V. R.; Zhang, H. C.; Gauthier, D. A.; Reddy, I.; Maryanoff, B. E. *Org. Lett.* **2003**, *5*, 4537; (h) Deaton, K. R.; Gin, M. S. *Org. Lett.* **2003**, *5*, 2477; (i) Sung, M. J.; Pang, J.-H.; Park, S. B.; Cha, J. K. *Org. Lett.* **2003**, *5*, 2137.
- Also see: (a) Boñaga, L. V. R.; Zhang, H. C.; Moretto, A. F.; Ye, H.; Gauthier, D. A.; Li, J.; Leo, G.; Maryanoff, B. E. *J. Am. Chem. Soc.* **2005**, *127*, 3473 and references therein; (b) Tanaka, K.; Nishida, G.; Ogino, M.; Hirano, M.; Noguchi, K. *Org. Lett.* **2005**, *7*, 3119; (c) Stará, I. G.; Alexandrová, Z.; Těplý, F.; Sehnal, P.; Stary, I.; Saman, D.; Budesinsky, M.; Cvacka, J. *Org. Lett.* **2005**, *7*, 2547; (d) Shibata, T.; Fujimoto, T.; Yokota, K.; Takagi, K. *J. Am. Chem. Soc.* **2004**, *126*, 8382; (e) Gutnov, A.; Heller, B.; Fischer, C.; Drexler, H.-J.; Spannenberg, A.; Sundermann, B.; Sundermann, C. *Angew. Chem., Int. Ed.* **2004**, *43*, 3795; (f) Tanaka, K.; Nishida, G.; Wada, A.; Noguchi, K. *Angew. Chem., Int. Ed.* **2004**, *43*, 6510; (g) Nishii, Y.; Wakasugi, K.; Koga, K.; Tanabe, Y. *J. Am. Chem. Soc.* **2004**, *126*, 5358.
- Tracey, M. R.; Zhang, Y.; Frederick, M. O.; Mulder, J. A.; Hsung, R. P. *Org. Lett.* **2004**, *6*, 2209.
- Tanaka, K.; Takeishi, K.; Noguchi, K. *J. Am. Chem. Soc.* **2006**, *128*, 4586.
- Tanaka, K.; Takeishi, K. *Synthesis* **2007**, 2920.
- For some elegant examples of axially chiral amido-aryl bonds: (a) Kitagawa, O.; Takahashi, M.; Yoshihara, M.; Taguchi, T. *J. Am. Chem. Soc.* **2005**, *127*, 3676; (b) Bennett, D. J.; Pickering, P. L.; Simpkins, N. S. *Chem. Commun.* **2004**, 1392; (c) Terauchi, J.; Curran, D. P. *Tetrahedron: Asymmetry* **2003**, *14*, 587; (d) Kitagawa, O.; Kohriyama, M.; Taguchi, T. *J. Org. Chem.* **2002**, *67*, 8682; (e) Ates, A.; Curran, D. P. *J. Am. Chem. Soc.* **2001**, *123*, 5130; (f) Hata, T.; Koide, H.; Taniguchi, N.; Uemura, M. *Org. Lett.* **2000**, *2*, 1907; (g) Shimizu, K. D.; Freyer, H. O.; Adams, R. D. *Tetrahedron Lett.* **2000**, *41*, 5431; (h) Kondo, K.; Fujita, H.; Suzuki, T.; Murakami, Y. *Tetrahedron Lett.* **1999**, *40*, 5577; (i) Ahmed, A.; Bragg, R. A.; Clayden, J.; Lai, L. W.; McCarthy, C.; Pink, J. H.; Westlund, N.; Yasin, S. A. *Tetrahedron* **1998**, *54*, 13277.
- For pioneering work, see: (a) Kitagawa, O.; Izawa, H.; Sato, K.; Dobashi, A.; Taguchi, T.; Shiro, M. *J. Org. Chem.* **1998**, *63*, 2634; (b) Curran, D. P.; Qi, H.; Geib, S. J.; DeMello, N. C. *J. Am. Chem. Soc.* **1994**, *116*, 3131.
- For leading reviews, see: (a) Clayden, J. *Chem. Soc. Rev.* **2009**, *38*, 817; (b) Jorgensen, K. A. *Angew. Chem., Int. Ed.* **2000**, *39*, 3558; (c) Clayden, J. *Angew. Chem., Int. Ed.* **1997**, *36*, 949.
- For an excellent book on concepts of asymmetric catalysis, see: Walsh, P. J.; Kozlowski, M. C. *Fundamentals of Asymmetric Catalysis*; University Science Books: Sausalito, CA, 2008.
- For leading examples, see: (a) Xie, X.; Phuan, P. W.; Kozlowski, M. C. *Angew. Chem., Int. Ed.* **2003**, *42*, 2168; (b) Shibata, T.; Fujimoto, T.; Yokota, K.; Takagi, K. *J. Am. Chem. Soc.* **2004**, *126*, 8382; (c) For a diastereoselective example, see: Fogel, L.; Hsung, R. P.; Wulff, W. D.; Sommer, R. D.; Rheingold, A. L. *J. Am. Chem. Soc.* **2001**, *123*, 5580.
- For elegant examples of achiral template based asymmetric catalysis, see: (a) Sibi, M. P.; Soeta, T. *J. Am. Chem. Soc.* **2007**, *129*, 4522; (b) Sibi, M. P.; Stanley, L. M.; Nie, X.; Venkatraman, L.; Liu, M.; Jasperse, C. P. *J. Am. Chem. Soc.* **2007**, *129*, 395; (c) Sibi, M. P.; Manyem, S.; Palencia, H. J. *Am. Chem. Soc.* **2006**, *128*, 13660; (d) Sibi, M. P.; Stanley, L. M.; Jasperse, C. P. *J. Am. Chem. Soc.* **2005**, *127*, 8277; (e) Sibi, M. P.; Zhang, R.; Manyem, S. *J. Am. Chem. Soc.* **2003**, *125*, 9306.
- For a review on synthesis of ynamides, see: Tracey, M. R.; Hsung, R. P.; Antoline, J. A.; Kurtz, K. C. M.; Shen, L.; Slafer, B. W.; Zhang, Y. In *Science of Synthesis*,

- Houben-Weyl Methods of Molecular Transformations*; Weinreb, Steve M., Ed.; Georg Thieme KG: Stuttgart, Germany, 2005; Chapter 21.4.
25. For synthesis of ynamides through Ullmann-type coupling, see: (a) Zhang, Y.; Hsung, R. P.; Tracey, M. R.; Kurtz, K. C. M.; Vera, E. L. *Org. Lett.* **2004**, *6*, 1151; (b) Dunetz, J. R.; Danheiser, R. L. *Org. Lett.* **2003**, *5*, 4011; (c) Frederick, M. O.; Mulder, J. A.; Tracey, M. R.; Hsung, R. P.; Huang, J.; Kurtz, K. C. M.; Shen, L.; Douglas, C. J. *J. Am. Chem. Soc.* **2003**, *125*, 2368; (d) Zhang, X.; Zhang, Y.; Huang, J.; Hsung, R. P.; Kurtz, K. C. M.; Oppenheimer, J.; Petersen, M. E.; Sagamanova, I. K.; Tracey, M. R. *J. Org. Chem.* **2006**, *71*, 4170; (e) Sagamanova, I. K.; Kurtz, K. C. M.; Hsung, R. P. *Org. Synth.* **2007**, *84*, 359; (f) Kohnen, A. L.; Dunetz, J. R.; Danheiser, R. L. *Org. Synth.* **2007**, *84*, 88; (g) Riddell, N.; Villeneuve, K.; Tam, W. *Org. Lett.* **2005**, *7*, 3681.
26. For a recent oxidative cross-coupling, see: Hamada, T.; Ye, X.; Stahl, S. S. *J. Am. Chem. Soc.* **2008**, *130*, 833.
27. Throughout the paper, the capital *M* and *P* denote C–C axial chirality while small *m* and *p* denote C–N axial chirality and are listed second.
28. CCDC for compounds **32-M,p**, **18-P,p**, and **34-M,p** are 665008, 665009, and 722530, respectively. CCDC 665008, 665009, and 722530 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request.cif.